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4-Chloro-5-[(5,5-dimethyl-4,5-dihydroisoxazol-3-yl)sulfonylmethyl]-3-methyl-1-(2,2,2-trifluoroethyl)-1*H*-pyrazole

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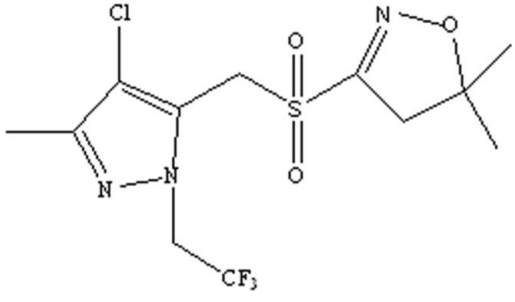
Key indicators: single-crystal X-ray study; T = 173 K; mean σ(C–C) = 0.003 Å; R factor = 0.040; wR factor = 0.097; data-to-parameter ratio = 13.6.

The molecule of the title compound, $C_{12}H_{15}ClF_3N_3O_3S$, is twisted, as indicated by the C—S—C—C torsion angle of 66.00 (18)° for the atoms linking the ring systems. An intramolecular C—H···F short contact occurs. In the crystal, non-classical C—H···O interactions, one of which has a short H···O contact of 2.28 Å, link the molecules.

Related literature

For background to pyrazoles and their pharmacological and pharmaceutical applications, see: Hirai *et al.* (2002); Shiga *et al.* (2003); Ohno *et al.* (2004); Sabbagh *et al.* (2009); Sridhar *et al.* (2004); Zheng *et al.* (2009).



Experimental

Crystal data

$C_{12}H_{15}ClF_3N_3O_3S$
M_r = 373.78
Monoclinic, $P2_1/n$
a = 16.034 (3) Å
b = 5.4319 (11) Å
c = 19.069 (4) Å
β = 106.71 (3)°
V = 1590.7 (6) Å^3
Z = 4
Cu $K\alpha$ radiation
μ = 3.83 mm^{-1}
T = 173 K
0.39 × 0.26 × 0.25 mm

Data collection

Rigaku R-AXIS RAPID IP diffractometer
Absorption correction: numerical (*NUMABS*; Higashi, 2003)
T_{min} = 0.317, T_{max} = 0.448
11348 measured reflections
2891 independent reflections
2568 reflections with $I > 2\sigma(I)$
R_{int} = 0.073

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.040
$wR(F^2)$ = 0.097
S = 1.09
2891 reflections
212 parameters
H-atom parameters constrained
$\Delta\rho_{max}$ = 0.39 e Å^{-3}
$\Delta\rho_{min}$ = −0.36 e Å^{-3}

Table 1
Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
C7—H7A···F1	0.99	2.44	3.229 (3)	136
C4—H4A···O3^{i}	0.98	2.60	3.325 (3)	131
C5—H5A···O2^{ii}	0.99	2.31	3.146 (3)	141
C7—H7B···O1^{iii}	0.99	2.28	3.265 (3)	171

Symmetry codes: (i) $-x+1, -y+1, -z$; (ii) $-x+\frac{1}{2}, y-\frac{1}{2}, -z+\frac{1}{2}$; (iii) $x, y+1, z$.

Data collection: *RAPID-AUTO* (Rigaku, 2001); cell refinement: *RAPID-AUTO*; data reduction: *RAPID-AUTO*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *Mercury* (Macrae *et al.*, 2006); software used to prepare material for publication: *SHELXL97*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB5103).

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supplementary materials

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4-Chloro-5-[(5,5-dimethyl-4,5-dihydroisoxazol-3-yl)sulfonylmethyl]-3-methyl-1-(2,2,2-trifluoroethyl)-1*H*-pyrazole

H.-J. Ma, Q.-F. Zhao, X.-D. Mei and J. Ning

Comment

Pyrazoles are an important class of compounds, which possess widespread pharmacological properties in pharmaceuticals (Sridhar *et al.*, 2004; Zheng *et al.*, 2009; Sabbagh *et al.*, 2009) and agrochemicals (Shiga *et al.*, 2003; Ohno *et al.*, 2004). Various pyrazole derivatives with potent herbicidal activity have been synthesized and some are in use as herbicides such as pyrazolate, pyrazoxyfen, benzofenap, pyraflufen-ethyl, fluazolate and pyrazosulfuron-ethyl (Hirai *et al.*, 2002). Recently, the new title compound (I) was synthesized in our group with high herbicidal activity. The crystal structure of the title compound is shown in Fig. 1.

Experimental

The title compound (0.2 g) was dissolved in acetone (50 ml) at room temperature. Colourless blocks of (I) were obtained through slow evaporation after two weeks.

Refinement

The H atoms were placed at calculated positions, with C—H = 0.93–0.98 Å, and refined as riding with U_{iso}(H) = 1.2–1.5U_{eq}(C).

Figures

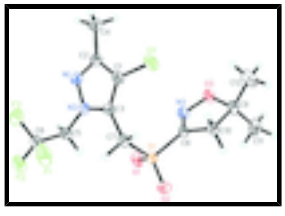


Fig. 1. The molecular structure of the title compound, showing 50% probability displacement ellipsoids and the atom-numbering scheme.

4-Chloro-5-[(5,5-dimethyl-4,5-dihydroisoxazol-3-yl)sulfonylmethyl]- 3-methyl-1-(2,2,2-trifluoroethyl)-1*H*-pyrazole

Crystal data

$C_{12}H_{15}ClF_3N_3O_3S$	F_{000} = 768
M_r = 373.78	D_x = 1.561 Mg m^{-3}
Monoclinic, $P2_1/n$	Cu $K\alpha$ radiation, λ = 1.54178 Å
Hall symbol: -P 2yn	Cell parameters from 11348 reflections
a = 16.034 (3) Å	θ = 3.2–68.2°

b = 5.4319 (11) Å	μ = 3.83 mm^{-1}
c = 19.069 (4) Å	T = 173 K
β = 106.71 (3)º	Block, colourless
V = 1590.7 (6) Å^3	0.39 × 0.26 × 0.25 mm
Z = 4	

Data collection

Rigaku R-AXIS RAPID IP diffractometer	2891 independent reflections
Radiation source: rotating anode	2568 reflections with $I > 2\sigma(I)$
Monochromator: graphite	R_{int} = 0.073
T = 173 K	θ_{max} = 68.2º
ω scans	θ_{min} = 3.2º
Absorption correction: numerical (NUMABS; Higashi, 2003)	h = −16→19
T_{min} = 0.317, T_{max} = 0.448	k = −6→5
11348 measured reflections	l = −22→20

Refinement

Refinement on F^2	Hydrogen site location: inferred from neighbouring sites
Least-squares matrix: full	H-atom parameters constrained
$R[F^2 > 2\sigma(F^2)]$ = 0.040	$w = 1/[\sigma^2(F_o^2) + (0.0203P)^2 + 1.0073P]$ where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2)$ = 0.097	$(\Delta/\sigma)_{max} < 0.001$
S = 1.09	$\Delta\rho_{max}$ = 0.39 e Å^{-3}
2891 reflections	$\Delta\rho_{min}$ = −0.36 e Å^{-3}
212 parameters	Extinction correction: SHELXL97 (Sheldrick, 2008), $Fc^*=kFc[1+0.001xFc^2\lambda^3/\sin(2\theta)]^{-1/4}$
Primary atom site location: structure-invariant direct methods	Extinction coefficient: 0.0054 (3)
Secondary atom site location: difference Fourier map	

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R-factor wR and goodness of fit S are based on F^2, conventional R-factors R are based on F, with F set to zero for negative F^2. The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R-factors(gt) *etc*. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F, and R- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å^2)

	x	*y*	*z*	U_{iso}*/U_{eq}
Cl1	0.44152 (4)	0.93466 (11)	0.09341 (4)	0.03950 (19)
S1	0.36881 (3)	0.46279 (10)	0.24281 (3)	0.02353 (17)
F1	0.10868 (10)	0.5476 (3)	0.06789 (10)	0.0610 (5)
F2	0.05756 (9)	0.1893 (3)	0.03398 (9)	0.0584 (5)
F3	0.09944 (9)	0.4206 (3)	−0.04032 (8)	0.0519 (5)
O1	0.35709 (10)	0.2075 (3)	0.22345 (9)	0.0320 (4)
O2	0.34961 (10)	0.5475 (3)	0.30760 (9)	0.0355 (4)
O3	0.60616 (9)	0.5020 (3)	0.23839 (9)	0.0295 (4)
N1	0.29777 (12)	0.3998 (4)	−0.01380 (10)	0.0308 (5)
N2	0.27450 (11)	0.4064 (3)	0.04952 (10)	0.0252 (4)
N3	0.52262 (11)	0.4070 (3)	0.22312 (10)	0.0273 (4)
C1	0.37126 (13)	0.6956 (4)	0.06043 (13)	0.0286 (5)
C2	0.35682 (14)	0.5758 (5)	−0.00755 (13)	0.0311 (5)
C3	0.31789 (13)	0.5852 (4)	0.09606 (12)	0.0238 (5)
C4	0.39748 (17)	0.6278 (6)	−0.06716 (14)	0.0478 (7)
H4A	0.3737	0.5148	−0.1081	0.072*
H4B	0.4606	0.6049	−0.0486	0.072*
H4C	0.3848	0.7979	−0.0840	0.072*
C5	0.20776 (14)	0.2398 (4)	0.05769 (12)	0.0277 (5)
H5A	0.2192	0.1968	0.1101	0.033*
H5B	0.2102	0.0861	0.0304	0.033*
C6	0.11822 (15)	0.3496 (5)	0.03000 (14)	0.0358 (6)
C7	0.30537 (13)	0.6436 (4)	0.16871 (12)	0.0252 (5)
H7A	0.2431	0.6211	0.1654	0.030*
H7B	0.3197	0.8193	0.1798	0.030*
C8	0.47791 (13)	0.5464 (4)	0.25218 (11)	0.0217 (5)
C9	0.61812 (13)	0.6994 (4)	0.29469 (12)	0.0262 (5)
C10	0.52374 (13)	0.7681 (4)	0.29159 (13)	0.0280 (5)
H10A	0.5161	0.7844	0.3411	0.034*
H10B	0.5045	0.9214	0.2635	0.034*
C11	0.66650 (17)	0.5831 (5)	0.36693 (14)	0.0415 (7)
H11A	0.7212	0.5121	0.3632	0.062*
H11B	0.6306	0.4531	0.3790	0.062*
H11C	0.6791	0.7087	0.4055	0.062*
C12	0.66893 (16)	0.9029 (5)	0.27166 (17)	0.0426 (7)
H12A	0.7249	0.8382	0.2688	0.064*
H12B	0.6791	1.0365	0.3077	0.064*
H12C	0.6357	0.9658	0.2236	0.064*

Atomic displacement parameters (Å^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Cl1	0.0318 (3)	0.0362 (4)	0.0491 (4)	−0.0112 (2)	0.0094 (3)	0.0061 (3)
S1	0.0209 (3)	0.0256 (3)	0.0244 (3)	−0.0035 (2)	0.0071 (2)	−0.0022 (2)

F1	0.0316 (8)	0.0694 (12)	0.0759 (13)	0.0110 (8)	0.0057 (8)	−0.0260 (10)
F2	0.0300 (8)	0.0797 (13)	0.0579 (10)	−0.0223 (8)	0.0005 (7)	0.0141 (9)
F3	0.0366 (8)	0.0723 (12)	0.0408 (9)	0.0071 (8)	0.0018 (7)	0.0211 (8)
O1	0.0305 (9)	0.0225 (9)	0.0404 (10)	−0.0064 (7)	0.0060 (7)	−0.0002 (7)
O2	0.0297 (9)	0.0528 (12)	0.0271 (9)	−0.0028 (8)	0.0132 (7)	−0.0060 (8)
O3	0.0204 (8)	0.0306 (9)	0.0379 (10)	−0.0020 (6)	0.0091 (7)	−0.0091 (7)
N1	0.0277 (10)	0.0423 (12)	0.0227 (10)	0.0014 (9)	0.0077 (8)	0.0003 (9)
N2	0.0234 (9)	0.0295 (10)	0.0227 (10)	−0.0022 (8)	0.0066 (8)	0.0004 (8)
N3	0.0219 (9)	0.0281 (10)	0.0308 (10)	−0.0018 (8)	0.0060 (8)	−0.0038 (8)
C1	0.0193 (11)	0.0307 (13)	0.0338 (13)	−0.0010 (9)	0.0047 (9)	0.0068 (10)
C2	0.0226 (11)	0.0424 (14)	0.0279 (13)	0.0016 (10)	0.0068 (10)	0.0081 (10)
C3	0.0207 (10)	0.0237 (11)	0.0262 (12)	0.0025 (8)	0.0053 (9)	0.0014 (9)
C4	0.0345 (14)	0.079 (2)	0.0333 (15)	−0.0023 (14)	0.0147 (11)	0.0113 (14)
C5	0.0269 (11)	0.0280 (12)	0.0261 (12)	−0.0059 (9)	0.0043 (9)	0.0015 (9)
C6	0.0242 (12)	0.0444 (15)	0.0368 (14)	−0.0086 (11)	0.0054 (10)	0.0017 (12)
C7	0.0212 (10)	0.0238 (11)	0.0309 (12)	0.0018 (9)	0.0080 (9)	−0.0027 (9)
C8	0.0198 (10)	0.0211 (11)	0.0231 (11)	0.0010 (8)	0.0043 (9)	−0.0008 (8)
C9	0.0221 (11)	0.0208 (11)	0.0335 (13)	−0.0012 (9)	0.0048 (9)	−0.0038 (9)
C10	0.0221 (11)	0.0248 (12)	0.0365 (13)	−0.0030 (9)	0.0076 (9)	−0.0079 (10)
C11	0.0348 (14)	0.0444 (16)	0.0367 (15)	0.0013 (11)	−0.0034 (11)	−0.0013 (12)
C12	0.0295 (13)	0.0306 (14)	0.070 (2)	−0.0011 (10)	0.0186 (13)	0.0039 (13)

Geometric parameters (Å, °)

Cl1—C1	1.716 (2)	C4—H4B	0.9800
S1—O2	1.4324 (16)	C4—H4C	0.9800
S1—O1	1.4333 (16)	C5—C6	1.503 (3)
S1—C8	1.766 (2)	C5—H5A	0.9900
S1—C7	1.780 (2)	C5—H5B	0.9900
F1—C6	1.329 (3)	C7—H7A	0.9900
F2—C6	1.324 (3)	C7—H7B	0.9900
F3—C6	1.344 (3)	C8—C10	1.496 (3)
O3—N3	1.386 (2)	C9—C12	1.511 (3)
O3—C9	1.490 (3)	C9—C11	1.511 (3)
N1—C2	1.327 (3)	C9—C10	1.543 (3)
N1—N2	1.363 (2)	C10—H10A	0.9900
N2—C3	1.364 (3)	C10—H10B	0.9900
N2—C5	1.444 (3)	C11—H11A	0.9800
N3—C8	1.274 (3)	C11—H11B	0.9800
C1—C3	1.374 (3)	C11—H11C	0.9800
C1—C2	1.409 (3)	C12—H12A	0.9800
C2—C4	1.491 (3)	C12—H12B	0.9800
C3—C7	1.490 (3)	C12—H12C	0.9800
C4—H4A	0.9800		
O2—S1—O1	119.23 (10)	F2—C6—C5	111.2 (2)
O2—S1—C8	106.40 (10)	F1—C6—C5	112.21 (19)
O1—S1—C8	109.12 (10)	F3—C6—C5	112.4 (2)
O2—S1—C7	106.96 (10)	C3—C7—S1	114.93 (15)
O1—S1—C7	109.07 (10)	C3—C7—H7A	108.5

C8—S1—C7	105.17 (10)	S1—C7—H7A	108.5
N3—O3—C9	109.60 (15)	C3—C7—H7B	108.5
C2—N1—N2	105.74 (19)	S1—C7—H7B	108.5
N1—N2—C3	112.16 (18)	H7A—C7—H7B	107.5
N1—N2—C5	118.62 (18)	N3—C8—C10	116.13 (19)
C3—N2—C5	129.16 (19)	N3—C8—S1	117.92 (16)
C8—N3—O3	108.56 (17)	C10—C8—S1	125.96 (16)
C3—C1—C2	107.0 (2)	O3—C9—C12	106.57 (19)
C3—C1—Cl1	125.86 (19)	O3—C9—C11	106.36 (18)
C2—C1—Cl1	127.13 (18)	C12—C9—C11	113.1 (2)
N1—C2—C1	109.9 (2)	O3—C9—C10	103.00 (16)
N1—C2—C4	121.7 (2)	C12—C9—C10	114.63 (19)
C1—C2—C4	128.5 (2)	C11—C9—C10	112.1 (2)
N2—C3—C1	105.2 (2)	C8—C10—C9	99.16 (17)
N2—C3—C7	125.2 (2)	C8—C10—H10A	111.9
C1—C3—C7	129.5 (2)	C9—C10—H10A	111.9
C2—C4—H4A	109.5	C8—C10—H10B	111.9
C2—C4—H4B	109.5	C9—C10—H10B	111.9
H4A—C4—H4B	109.5	H10A—C10—H10B	109.6
C2—C4—H4C	109.5	C9—C11—H11A	109.5
H4A—C4—H4C	109.5	C9—C11—H11B	109.5
H4B—C4—H4C	109.5	H11A—C11—H11B	109.5
N2—C5—C6	112.07 (19)	C9—C11—H11C	109.5
N2—C5—H5A	109.2	H11A—C11—H11C	109.5
C6—C5—H5A	109.2	H11B—C11—H11C	109.5
N2—C5—H5B	109.2	C9—C12—H12A	109.5
C6—C5—H5B	109.2	C9—C12—H12B	109.5
H5A—C5—H5B	107.9	H12A—C12—H12B	109.5
F2—C6—F1	107.5 (2)	C9—C12—H12C	109.5
F2—C6—F3	106.76 (19)	H12A—C12—H12C	109.5
F1—C6—F3	106.4 (2)	H12B—C12—H12C	109.5
C2—N1—N2—C3	0.1 (2)	N2—C3—C7—S1	86.0 (2)
C2—N1—N2—C5	177.64 (19)	C1—C3—C7—S1	−95.8 (3)
C9—O3—N3—C8	11.2 (2)	O2—S1—C7—C3	178.87 (16)
N2—N1—C2—C1	0.0 (2)	O1—S1—C7—C3	−50.91 (19)
N2—N1—C2—C4	−179.3 (2)	C8—S1—C7—C3	66.00 (18)
C3—C1—C2—N1	−0.2 (3)	O3—N3—C8—C10	1.4 (3)
Cl1—C1—C2—N1	180.00 (17)	O3—N3—C8—S1	−178.52 (13)
C3—C1—C2—C4	179.0 (2)	O2—S1—C8—N3	148.32 (18)
Cl1—C1—C2—C4	−0.8 (4)	O1—S1—C8—N3	18.5 (2)
N1—N2—C3—C1	−0.2 (2)	C7—S1—C8—N3	−98.41 (19)
C5—N2—C3—C1	−177.4 (2)	O2—S1—C8—C10	−31.6 (2)
N1—N2—C3—C7	178.34 (19)	O1—S1—C8—C10	−161.43 (18)
C5—N2—C3—C7	1.2 (3)	C7—S1—C8—C10	81.7 (2)
C2—C1—C3—N2	0.2 (2)	N3—O3—C9—C12	−139.32 (18)
Cl1—C1—C3—N2	−179.92 (16)	N3—O3—C9—C11	99.8 (2)
C2—C1—C3—C7	−178.3 (2)	N3—O3—C9—C10	−18.3 (2)
Cl1—C1—C3—C7	1.6 (3)	N3—C8—C10—C9	−12.4 (3)
N1—N2—C5—C6	−89.8 (2)	S1—C8—C10—C9	167.48 (16)

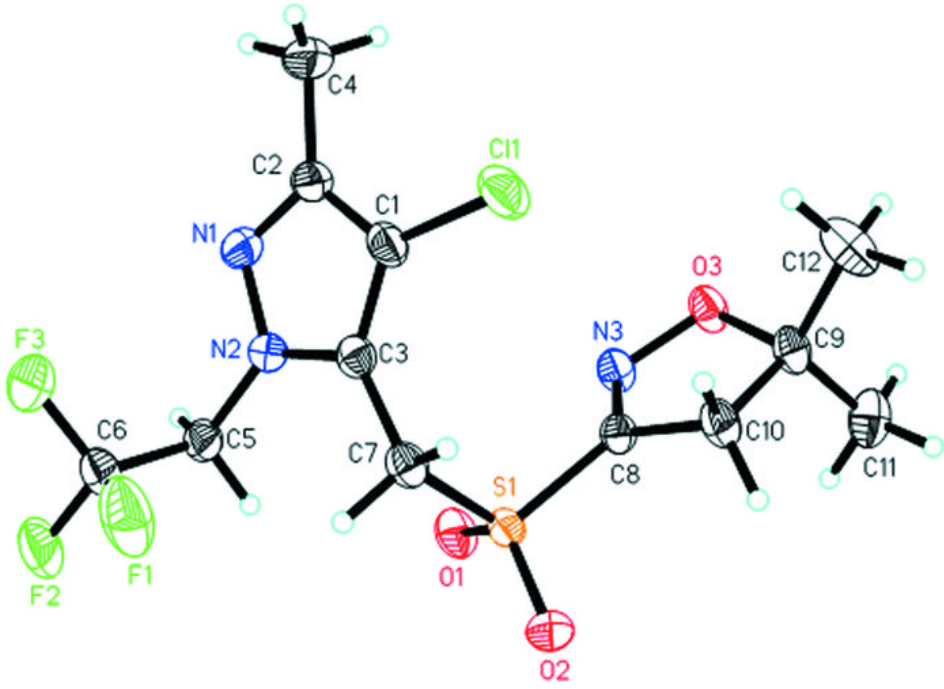
C3—N2—C5—C6	87.3 (3)	O3—C9—C10—C8	17.0 (2)
N2—C5—C6—F2	176.34 (19)	C12—C9—C10—C8	132.3 (2)
N2—C5—C6—F1	−63.3 (3)	C11—C9—C10—C8	−96.9 (2)
N2—C5—C6—F3	56.7 (3)		

Hydrogen-bond geometry (Å, °)

D—H···*A*	*D*—H	H···*A*	*D*···*A*	*D*—H···*A*
C7—H7A···F1	0.99	2.44	3.229 (3)	136
C4—H4A···O3^{i}	0.98	2.60	3.325 (3)	131
C5—H5A···O2^{ii}	0.99	2.31	3.146 (3)	141
C7—H7B···O1^{iii}	0.99	2.28	3.265 (3)	171

Symmetry codes: (i) −*x*+1, −*y*+1, −*z*; (ii) −*x*+1/2, *y*−1/2, −*z*+1/2; (iii) *x*, *y*+1, *z*.

Fig. 1


C4
C2
Cl1
C1
N1
O3
C12
F3
N2
C3
N3
C9
C6
C5
C7
S1
C8
C10
C11
F2
F1
O1
O2